Molar Conversion WS

Calculate how many moles are in the following masses: 1) 25 g of NaCl 2) 125 g of H₂SO₄ Calculate the mass (in grams) of the following #of moles: 3) 2.5 mol of NaCl 4) 0.5 mole of H₂SO₄ How many molecules are in the following number of moles? 5) 2 moles of NaCl 6) 1.5 moles H₂SO₄ How many moles are in the following # of molecules? 7) 3.4 x 10²⁶ of NaCl 8) 7.5 x 10¹⁹ of H₂SO₄ How many molecules are in the following # of grams? 9) 87 g of NaCl 10) 45 g of H₂SO₄ How many grams are in the following # of molecules? 11) 1.8 x 10²⁸ of NaCl 12) 4.5 x 10¹⁵ of H₂SO₄

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